

**University of Calcutta under Graduate Curriculum under Choice Based Credit System (CBCS)**

**Lesson Plan with Syllabus for Chemistry (G) Semester-IV**

**Total Marks-100 (Credits: Theory-04, Practical-02)**

**(Theory: 50; Practical: 30; Internal Assessment: 10; Attendance: 10)**

**[Marks obtained in this course will be taken to calculate SGPA & CGPA]**

Tentative may subject to change:					
Months	Week	Unit	Topic	No. of Lectures for Each Section	Teacher
February	3 <sup>rd</sup>	1	<p align="center"><b>Alcohols, Phenols, and Ethers</b></p> <p align="center"><i>Alcohols</i></p> <ul style="list-style-type: none"> <li>Preparation: 1<sup>o</sup>-, 2<sup>o</sup>- and 3<sup>o</sup>- alcohols: using a Grignard reagent, reduction of aldehydes, ketones, carboxylic acid, and esters</li> </ul>	2	TKL
		4	<p align="center"><b>Amines and Diazonium Salt</b></p> <ul style="list-style-type: none"> <li>Amines: strength of organic bases; Preparation: from alkyl halides, Hofmann degradation</li> </ul>	1	SM
		6	<p align="center"><b>Crystal Field Theory (CFT)</b></p> <ul style="list-style-type: none"> <li>Postulates of Crystal field theory</li> <li>Crystal field splitting of Octahedral complex and effects for weak and strong field ligand</li> </ul>	1	PKD
March	1 <sup>st</sup>	1	<p align="center"><b>Alcohols, Phenols, and Ethers</b></p> <p align="center"><i>Alcohols</i></p> <ul style="list-style-type: none"> <li>Reactions: With sodium, oxidation (alkaline KMnO<sub>4</sub>, acidic dichromate)</li> </ul>	2	TKL
		4	<p align="center"><b>Amines and Diazonium Salt</b></p> <ul style="list-style-type: none"> <li>Reactions: with HNO<sub>2</sub> (distinction of 1<sup>o</sup>-, 2<sup>o</sup>- and 3<sup>o</sup>- amines), Schotten- Baumann reaction</li> </ul>	1	SM
		6	<p align="center"><b>Crystal Field Theory (CFT)</b></p> <ul style="list-style-type: none"> <li>Crystal field splitting of Tetrahedral complex and factors affecting the magnitude of D</li> <li>Comparison of CFSE for Octahedral and Tetrahedral complexes, Spectrochemical series</li> </ul>	1	PKD
March	2 <sup>nd</sup>	1	<p align="center"><b>Alcohols, Phenols, and Ethers</b></p> <p align="center"><i>Alcohols</i></p> <ul style="list-style-type: none"> <li>Diols: Pinacol-pinacolone rearrangement (with mechanism) (with symmetrical diols only)</li> </ul>	2	TKL
		4	<p align="center"><b>Amines and Diazonium Salt</b></p> <ul style="list-style-type: none"> <li>Diazo coupling reaction (with mechanism)</li> </ul>	1	SM
		6	<p align="center"><b>Crystal Field Theory (CFT)</b></p> <ul style="list-style-type: none"> <li>Tetragonal distortion of Octahedral geometry, Jahn-Teller distortion, Square planar geometry</li> </ul>	1	PKD
	3 <sup>rd</sup>	1	<p align="center"><b>Alcohols, Phenols, and Ethers</b></p> <p align="center"><i>Phenols</i></p> <ul style="list-style-type: none"> <li>Preparation: cumene hydroperoxide method, from diazonium salts; acidic nature of phenols</li> </ul>	2	TKL
		4	<p align="center"><b>Amines and Diazonium Salt</b></p> <ul style="list-style-type: none"> <li>Diazonium salts: Preparation: from aromatic amines; Reactions: conversion to benzene, phenol, benzoic acid, and nitrobenzene</li> </ul>	1	SM
		7	<p align="center"><b>Quantum Chemistry and Spectroscopy</b></p> <ul style="list-style-type: none"> <li>Basic Concept of Electromagnetic radiation and its interaction with matter</li> </ul>	1	PKD
	4 <sup>th</sup>	1	<p align="center"><b>Alcohols, Phenols, and Ethers</b></p> <p align="center"><i>Phenols</i></p> <ul style="list-style-type: none"> <li>Reactions: electrophilic substitution: nitration and halogenations; Reimer-Tiemann reaction,</li> </ul>	2	TKL

			Schotten Baumann reaction, Fries rearrangement, and Claisen rearrangement		
		4	<p style="text-align: center;"><b>Amines and Diazonium Salt</b></p> <ul style="list-style-type: none"> <li>Nitro compounds (aromatic): reduction under different conditions (acidic, neutral, and alkaline)</li> </ul>	1	SM
		7	<p style="text-align: center;"><b>Quantum Chemistry and Spectroscopy</b></p> Wave-particle duality, the link between spectroscopy and quantum chemistry	1	TKL
	5 <sup>th</sup>		<b>**Student Lecture: On Crystal Field Theory (CFT)</b>	1	PKD, SM, TKL

Months	Week	Unit	Topic	No. of Lectures for Each Section	Teacher
April	1 <sup>st</sup>	1	<p style="text-align: center;"><b>Alcohols, Phenols, and Ethers</b></p> <p style="text-align: center;"><i>Ethers</i></p> <ul style="list-style-type: none"> <li>Preparation: Williamson's ether synthesis;</li> <li>Reaction: cleavage of ethers with HI</li> </ul>	2	TKL
		5	<p style="text-align: center;"><b>Amino Acids and Carbohydrates</b></p> <p style="text-align: center;"><i>Amino Acids</i></p> <ul style="list-style-type: none"> <li>Preparations (glycine and alanine only): Strecker synthesis, Gabriel's phthalimide</li> </ul>	1	SM
		7	<p style="text-align: center;"><b>Quantum Chemistry and Spectroscopy</b></p> <ul style="list-style-type: none"> <li>Spectroscopy and its importance in chemistry</li> <li>types of spectroscopy</li> </ul>	1	PKD
	2 <sup>nd</sup>	2	<p style="text-align: center;"><b>Carbonyl Compounds</b></p> <p style="text-align: center;"><i>Aldehydes and Ketones</i></p> <ul style="list-style-type: none"> <li>Preparation: from acid chlorides, nitriles, and Grignard reagents; general properties of aldehydes and ketones</li> </ul>	2	TKL
		5	<p style="text-align: center;"><b>Amino Acids and Carbohydrates</b></p> <p style="text-align: center;"><i>Amino Acids</i></p> <ul style="list-style-type: none"> <li>Synthesis; general properties; zwitterion, isoelectric point</li> </ul>	1	SM
		7	<p style="text-align: center;"><b>Quantum Chemistry and Spectroscopy</b></p> <ul style="list-style-type: none"> <li>Difference between atomic and molecular spectra</li> <li>Postulates of quantum mechanics and quantum mechanical operator</li> </ul>	1	PKD
			McQ based Assessment for all 3-section on Unit-1, 4 & 6	1	PKD, SM, TKL
	3 <sup>rd</sup>	2	<p style="text-align: center;"><b>Carbonyl Compounds</b></p> <p style="text-align: center;"><i>Aldehydes and Ketones</i></p> <ul style="list-style-type: none"> <li>Reactions: with HCN, NaHSO<sub>3</sub>, NH<sub>2</sub>-G derivatives and with Tollens' and Fehling's reagents; iodoform test; aldol condensation (with mechanism)</li> </ul>	2	TKL
		5	<p style="text-align: center;"><b>Amino Acids and Carbohydrates</b></p> <p style="text-align: center;"><i>Carbohydrates</i></p> <ul style="list-style-type: none"> <li>classification and general properties; glucose and fructose: constitution</li> </ul>	1	SM

		7	<b><i>Quantum Chemistry and Spectroscopy</i></b> <ul style="list-style-type: none"> <li>Free particle, Particle in a 1-D box</li> </ul>	1	PKD
	4 <sup>th</sup>	2	<b><i>Carbonyl Compounds</i></b> <i>Aldehydes and Ketones</i> <ul style="list-style-type: none"> <li>Cannizzaro reaction (with mechanism), Wittig reaction, benzoin condensation; Clemmensen reduction, Wolff-Kishner reduction</li> </ul>	2	TKL
		5	<b><i>Amino Acids and Carbohydrates</i></b> <i>Carbohydrates</i> <ul style="list-style-type: none"> <li>Osazone formation; oxidation-reduction reactions</li> </ul>	1	SM
		7	<b><i>Quantum Chemistry and Spectroscopy</i></b> <ul style="list-style-type: none"> <li>Normalization of wave functions and concept of zero-point energy</li> <li>Rotational Motion: Schrodinger equation of a rigid rotator and its results</li> </ul>	1	PKD
May	1 <sup>st</sup>		**** <b><i>Library work assignment</i></b>		PKD, TKL, SM
	2 <sup>nd</sup>	3	<b><i>Carboxylic acid and their derivatives</i></b> <i>Carboxylic acids</i> <ul style="list-style-type: none"> <li>Strength of organic acids: a comparative study with emphasis on factors affecting pK values</li> <li><i>Preparation:</i> acidic and alkaline hydrolysis of esters (<math>B_{Ac2}</math> and <math>A_{Ac2}</math> mechanisms only) and from Grignard reagents</li> </ul>	2	TKL
		5	<b><i>Amino Acids and Carbohydrates</i></b> <i>Carbohydrates</i> <ul style="list-style-type: none"> <li>Ascending (Kiliani-Fischer method) and descending (Ruff's method) in monosaccharides (aldoses only)</li> </ul>	1	SM
		7	<b><i>Quantum Chemistry and Spectroscopy</i></b> <ul style="list-style-type: none"> <li>Quantization of rotational energy level</li> <li>Microwave spectra of diatomic molecules and selection rules</li> </ul>	1	PKD
	3 <sup>rd</sup>	3	<b><i>Carboxylic acid and their derivatives</i></b> <i>Carboxylic acids derivatives</i> <ul style="list-style-type: none"> <li><i>Preparation:</i> acid chlorides, anhydrides, esters, and amides from acids</li> <li><i>Reactions:</i> Interconversion among acid derivatives</li> </ul>	2	TKL
		5	<b><i>Amino Acids and Carbohydrates</i></b> <ul style="list-style-type: none"> <li>Mutarotation</li> </ul>	1	SM
		7	<b><i>Quantum Chemistry and Spectroscopy</i></b> <ul style="list-style-type: none"> <li>Structural information derived from rotational spectroscopy</li> </ul>	1	PKD
	4 <sup>th</sup>	3	<b><i>Carboxylic acid and their derivatives</i></b> <i>Carboxylic acids derivatives</i> <ul style="list-style-type: none"> <li><i>Reactions:</i> Claisen condensation</li> </ul>	2	TKL
		7	<b><i>Quantum Chemistry and Spectroscopy</i></b> <ul style="list-style-type: none"> <li>Vibrational Motion: Schrodinger equation of a linear harmonic oscillator and its results</li> </ul>	1	PKD

June	1 <sup>st</sup>		<b>**Guest Lecture</b>		
	2 <sup>nd</sup>	<b>3</b>	<b><i>Carboxylic acid and their derivatives</i></b> <ul style="list-style-type: none"> <li>▪ <i>Reactions:</i> Perkin reaction</li> <li>▪ Question answers discussion</li> </ul>	2	TKL
<b>7</b>		<b><i>Quantum Chemistry and Spectroscopy</i></b> <ul style="list-style-type: none"> <li>▪ Quantization of vibrational energy levels, selection rules</li> </ul>	1	PKD	
		<ul style="list-style-type: none"> <li>▪ Question answers discussion</li> </ul>	1	SM	
<b>Internal Assessment</b>		<b>McQ based Internal Assessment for all sections</b>		PKD, SM, TKL	

<u><i>Tentative may subject to change: Practical Class</i></u>			
Months	Weeks	Topic	Teacher
September	3 <sup>rd</sup>	<ul style="list-style-type: none"> <li>▪ Laboratory work discussion</li> </ul>	TKL
	4 <sup>th</sup> to 5 <sup>th</sup>	<ul style="list-style-type: none"> <li>• <b>Qualitative Analysis of Single Solid Organic Compound(s)</b></li> <li>• Experiment A: Detection of special elements (N, Cl, and S) in organic compounds</li> </ul>	
October	1 <sup>st</sup> to 2 <sup>nd</sup>	<ul style="list-style-type: none"> <li>• Experiment B: Solubility and Classification (solvents: H<sub>2</sub>O, dil. HCl, dil. NaOH)</li> </ul>	
November	1 <sup>st</sup> to 2 <sup>nd</sup>	<ul style="list-style-type: none"> <li>• Experiment C: Detection of functional groups: Aromatic -NO<sub>2</sub>, Aromatic -NH<sub>2</sub>, -COOH</li> </ul>	
	3 <sup>rd</sup> to 4 <sup>th</sup>	<ul style="list-style-type: none"> <li>• Experiment C: Carbonyl (no distinction of –CHO and &gt;C=O needed), -OH (phenolic) in solid organic compounds</li> </ul>	
December	1 <sup>st</sup> to 2 <sup>nd</sup>	<ul style="list-style-type: none"> <li>• Experiments A - C with unknown (at least 6) solid samples containing not more than two of the above type of functional groups should be done</li> </ul>	
	3 <sup>rd</sup> to 4 <sup>th</sup>	<ul style="list-style-type: none"> <li>• Experiments A - C with unknown (at least 6) solid samples containing not more than two of the above type of functional groups should be done</li> </ul>	
January	1 <sup>st</sup> to 2 <sup>nd</sup>	<p><b><i>Identification of a pure organic compound</i></b></p> <ul style="list-style-type: none"> <li>• <i>Solid compounds:</i> oxalic acid, tartaric acid, succinic acid, resorcinol, urea, glucose, benzoic acid and salicylic acid</li> </ul>	
	3 <sup>rd</sup> to 4 <sup>th</sup>	<ul style="list-style-type: none"> <li>• <i>Liquid Compounds:</i> methyl alcohol, ethyl alcohol, acetone, aniline, dimethylaniline, benzaldehyde, chloroform and nitrobenzene</li> </ul>	